

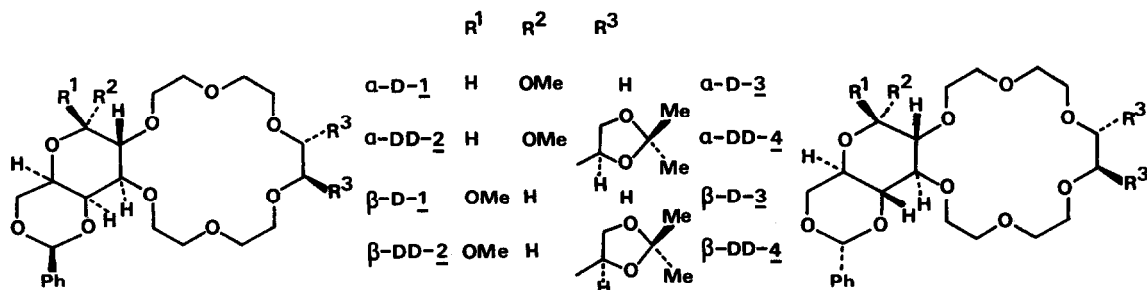
COMPLEXATION SELECTIVITY BY CHIRAL ASYMMETRIC CROWNS INCORPORATING THE 4,6-O-BENZYLIDENE DERIVATIVES OF METHYL β -D-GLUCOPYRANOSIDE AND METHYL β -D-GALACTOPYRANOSIDE. A SECONDARY ANOMERIC EFFECT?

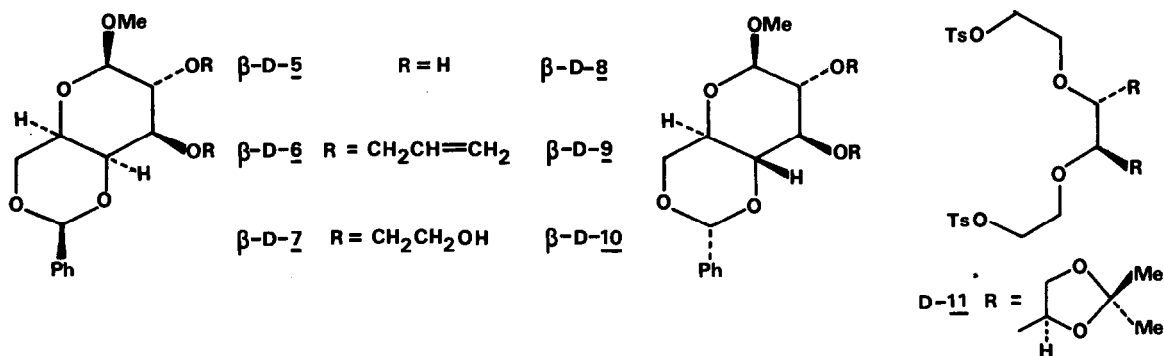
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Chiral asymmetric crowns have been derived from the 4,6-O-benzylidene derivatives of methyl α -D-gluco-,^{2a-c} α -D-galacto-,^{2a-c} and α -D-manno-^{2c,d} pyranosides. The temperature dependent ¹H n.m.r. spectra of the 1:1 complexes formed between the crowns α -D-(1) to α -DD-(4) and $\text{RNH}_3^+\text{SCN}^-$ salts in CD_2Cl_2 have been interpreted^{2b} in terms of equilibria involving anisometric⁶ α - and β -complexes of similar strengths in the case of the α -D-galacto crowns α -D-(1) and α -DD-(2) where both O-1 (α -complex) and O-4 (β -complex) on the pyranosidic ring participate^{2b} along with the six crown oxygens in complex formation. With the α -D-gluco crowns α -D-(3) and α -DD-(4), where only O-1 (α -complex) can participate along with the six crown oxygens in complex formation, anisometric³ α - and β -complexes of dissimilar strengths form with RNH_3^+ ions which contain a phenyl group with the appropriate constitutional disposition within R relative to the NH_3^+ ionic centre. The high selectivity in the complexation of $\text{PhCH}_2\text{NH}_3^+$ and PhCHMeNH_3^+ ions by the heterotopic faces of α -D-3 and α -DD-4 has been ascribed^{2b} to a dipole-induced dipole interaction⁴ between the 2-phenyl-1,3-dioxan ring in these crowns and the phenyl groups in the RNH_3^+ ions. As a result of this secondary interaction and the involvement of the axial O-1 in primary binding of the NH_3^+ ionic centre, it was of interest to explore the effect of inverting configuration at C-1 in α -D-1 to α -DD-4 on the selectivity of complexation of RNH_3^+ ions. Consequently, the anomers β -D-1 to β -DD-4 have been prepared and their 1:1 complexes in CD_2Cl_2 with selected $\text{RNH}_3^+\text{SCN}^-$ and $\text{RNH}_3^+\text{ClO}_4^-$ salts have now been studied by dynamic ¹H n.m.r. spectroscopy.

Treatment of methyl 4,6-O-benzylidene- β -D-galactopyranoside⁵ β -D-(5) with NaH and 1.1 molar equivs. of $\text{Ts}(\text{OCH}_2\text{CH}_2)_5\text{OTs}$ ⁶ in DMSO gave the β -D-galacto-18-crown-6 derivative β -D-(1), m.p. 141-143°, $[\alpha]_D + 30.2^\circ$ (c 0.5, CHCl_3)⁷ in 20% yield after chromatography on alumina (ether-methanol). By following a similar procedure, the β -D-gluco-18-crown-6 derivative β -D-(3), m.p. 78-80°, $[\alpha]_D - 47.0^\circ$ (c 0.5, CHCl_3)⁸ was obtained from methyl 4,6-O-benzylidene- β -D-glucopyranoside⁹ β -D-(8) in 46% yield after chromatography on alumina (ether). As in the





α -series^{2a,b} [cf. α -DD-2 and α -DD-4], the "half crown" bistosylate¹⁰ D-11 derived from D-mannitol was condensed with the "half crown" diols β -D-7 and β -D-10 obtained from the reaction sequences β -D-5 \rightarrow β -D-6 \rightarrow β -D-7¹¹ and β -D-8 \rightarrow β -D-9 \rightarrow β -D-10¹² to afford, respectively, the β -D-galacto-manno-¹³ β -DD-2, $[\alpha]_D + 26.5^\circ$ (c 0.5, CHCl₃)¹⁴ and the β -D-gluco-manno-¹⁵ β -DD-4, $[\alpha]_D - 98.4^\circ$ (c 0.5, CHCl₃)¹⁶ 18-crown-6 derivatives.

The association constants (K_a) for complexing of β -D-1, β -DD-2, β -D-3, and β -DD-4 with Me₃CNH₃⁺SCN⁻ in CDCl₃ were estimated¹⁷ to be 5800, 4200, 1300, and 1000 M⁻¹, respectively.¹⁸ In CD₂Cl₂, β -D-1 to β -DD-4 form 1:1 complexes with selected SCN⁻ and ClO₄⁻ salts derived from Me₃CNH₂ (12), PhCH₂NH₂ (13), and (*R*)- and (*S*)-PhCHMeNH₂ (*R*)- and (*S*)-14). Complex formation is accompanied by substantial changes in the ¹H n.m.r. spectra of all these crowns. The spectral behaviour of the β -D-1-13.HSCN complex serves to illustrate the general situation which pertains for the β -galacto crowns. At +30^o, the signals for the benzylidene CH (δ 5.62), H-1 (δ 4.39), and H-4 (δ 4.52) are shifted markedly downfield compared with their respective chemical shifts⁷ in the spectrum of β -D-1. This observation indicates that O-4 participates along with the six crown oxygens in complex formation. On cooling down to -90^o, the signals for the benzylidene CH and H-4 each separate into high (δ 5.63 and 4.56, respectively) and low (δ 5.76 and 4.75, respectively) intensity signals. The fact that the lower intensity signals in each case correspond to the lower field signals suggests that the minor complex is associated with the β -face of β -D-1. Thus, the major complex appears, rather unexpectedly, to involve binding to the α -face of β -D-1. The Table shows the generality of this conclusion for complexes of β -D-1 with 13.HX, (*R*)-14.HX, and (*S*)-14.HX and also lists the relative populations of the two complexes and the free energies of activation (ΔG^\ddagger) for dissociation of the major complexes in both the *galacto* and *gluco* series. The spectra of the 1:1 complexes involving

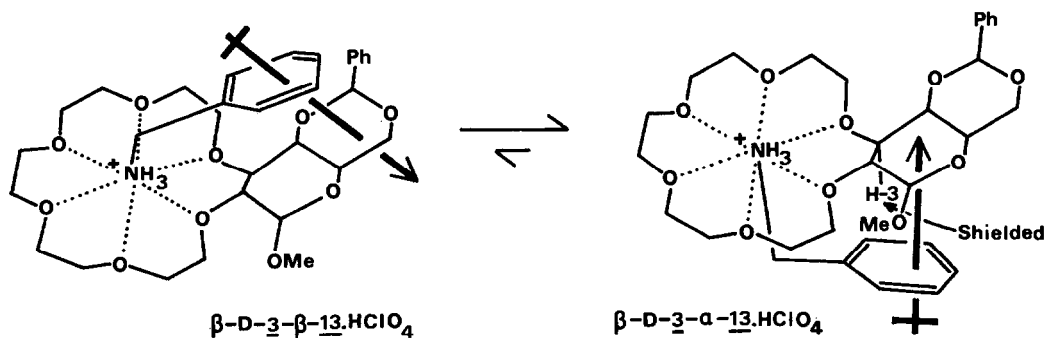


Table. Temperature dependent ^1H n.m.r. spectral data and thermodynamic parameters for the 1:1 complexation of selected RNH_3^+X^- salts 12.HSCN, 13.HSCN, 13.HClO₄, (*R*)-14.HSCN, (*S*)-14.HSCN, and (*S*)-14.HClO₄ by the chiral asymmetric crowns β -D-1, β -DD-2, β -D-3, and β -DD-4^a

Crown	RNH_3^+X^-	R	Downfield shift (p.p.m.) of benzyldiene CH at +30°	δ Benzyldiene CH at -90°		Complex ratio (T/°C) Maj : Min	ΔG_d^\ddagger (Maj \rightarrow Min) ± 0.3 kcal mol ⁻¹
β - <u>D</u> - <u>1</u>	<u>12</u> .HSCN	CMe ₃	0.12	-	-	76 : 24(-100)	10.0
	<u>13</u> .HSCN	CH ₂ Ph	0.11	5.63	5.76	78 : 22(-90)	12.0
	<u>13</u> .HClO ₄	CH ₂ Ph	0.08	5.58	5.70	78 : 22(-70)	10.8
	(<i>R</i>)- <u>14</u> .HSCN	CHMePh	0.10	5.59	5.72	70 : 30(-90)	10.5
	(<i>S</i>)- <u>14</u> .HSCN	CHMePh	0.11	5.65	5.75	87 : 13(-80)	11.7
	(<i>S</i>)- <u>14</u> .HClO ₄	CHMePh	0.08	5.57	5.70	73 : 27(-90)	10.6
β - <u>DD</u> - <u>2</u>	<u>12</u> .HSCN	CMe ₃	0.11	-	-	-	-
	<u>13</u> .HSCN	CH ₂ Ph	0.12	-	-	-	-
	(<i>S</i>)- <u>14</u> .HSCN	CHMePh	0.12	-	-	-	-
β - <u>D</u> - <u>3</u>	<u>12</u> .HSCN	CMe ₃	0.05	-	-	-	-
	<u>13</u> .HSCN	CH ₂ Ph	0.03	5.54	5.69	78 : 22(-90)	10.2
	<u>13</u> .HClO ₄	CH ₂ Ph	0.02	5.54	5.68	68 : 32(-90)	10.6
	(<i>R</i>)- <u>14</u> .HSCN	CHMePh	0.05	5.58	5.66	76 : 24(-100)	10.3
	(<i>S</i>)- <u>14</u> .HSCN	CHMePh	0.02	5.59	5.69	68 : 32(-80)	9.7
	(<i>S</i>)- <u>14</u> .HClO ₄	CHMePh	0.02	5.57	5.65	76 : 24(-90)	10.2
β - <u>DD</u> - <u>4</u>	<u>13</u> .HSCN	CH ₂ Ph	0.03	5.62	5.74	86 : 14(-90)	10.8
	(<i>S</i>)- <u>14</u> .HSCN	CHMePh	0.03	5.60	5.69	72 : 28(-90)	10.6

^aAll spectra were recorded in CD₂Cl₂ at 220 MHz on a Perkin Elmer R34 spectrometer with Me₄Si as "lock" and internal standard.

^bThe free energies of activation (ΔG_d^\ddagger) for dissociation of the 1:1 complexes correspond to the average values calculated from the Eyring equation using rate constants determined by line shape analysis of the benzyldiene CH signal at two different temperatures in each case.

β -DD-2 did not provide suitable temperature dependent behaviour in their potential ^1H n.m.r. probes and so did not afford thermodynamic data. The spectral behaviour of the β -D-3-13.HClO₄ complex serves to illustrate the general situation which pertains for the β -D-*gluco* crowns. Although the influence upon the chemical shift of the benzyldiene CH on complex formation is much less marked than in the *galacto* series, this ^1H n.m.r. probe shows temperature dependence presumably as a result of the dipole-induced dipole interaction present in the β -D-3- β -13.HClO₄ complex which no doubt brings the benzyldiene CH under the anisotropic influence of the phenyl ring. At -90°, a triplet ($J = 10$ Hz) with a peak area corresponding to ca. 0.8 of a proton emerged at high field (δ 2.62) and could be assigned to H-3 in the major complex on the basis of double irradiation studies. Since this shielding of H-3 is characteristic of nearly all the low temperature spectra of 1:1 complexes involving β -D-1 to β -DD-4 and phenyl-containing RNH_3^+ ions, we are led to propose the existence of a stabilising secondary anomeric effect in the α -complexes (*e.g.* β -D-3- α -13.HClO₄) which involves the dipole associated with the anomeric region of β -glycosides¹⁹ and an induced dipole in the phenyl ring of appropriate RNH_3^+ ions. It is intriguing to reflect on the possibility that the anomeric effect which destabilises²⁰ β -glycosides intramolecularly may be a source of stability "intermolecularly" within a complex of a β -glycoside!

References and Footnotes

1. Address all correspondence to this author at the Corporate Laboratory, Imperial Chemical Industries Ltd., P.O. Box No. 11, The Heath, Runcorn, Cheshire WA7 4QE.
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3. The term anisometric (K. Mislow, *Bull. Soc. Chim. Belg.*, 86, 595 (1977)) implies that the complexes are constitutionally isomeric as well as diastereoisomeric.
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7. ^1H N.m.r. data (CD_2Cl_2) for $\beta\text{-D-1}$: δ 7.55-7.30 (m, 5H, aromatic protons), 5.52 (s, 1H, benzyldiene CH), and 4.32-3.32 (m, 27H, other CH and CH_2 protons; s at 3.54, 3H, OCH_3).
8. ^1H N.m.r. data (CD_2Cl_2) for $\beta\text{-D-3}$: δ 7.50-7.30 (m, 5H, aromatic protons), 5.52 (s, 1H, benzyldiene CH), 4.37-4.26 (d, $J = 8$ Hz, 1H, H-1; q, $J = 5, 10$ Hz, 1H, H-6e), 3.95-3.45 (m, 25H, H-3, H-4, H-6e, and CH_2 protons; s at 3.50, 3H, OCH_3), 3.45-3.25 (m, 1H, H-5), and 3.15 (t, $J = 8$ Hz, 1H, H-2). The above assignments were made as a result of homonuclear INDOR spectroscopy as well as by conventional double irradiation experiments.
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11. The diol $\beta\text{-D-5}$, when treated with an excess of $\text{CH}_2=\text{CHCH}_2\text{Br}$ and KOH in toluene, gave the diallyl ether $\beta\text{-D-6}$, m.p. 131-132°, $[\alpha]_{\text{D}} + 32.7^\circ$ (c 0.5, CHCl_3) in 73% yield. Ozonolysis of $\beta\text{-D-6}$ in methanol, followed by borohydride reduction, afforded (74%) $\beta\text{-D-7}$, m.p. 150-151°, $[\alpha]_{\text{D}} + 32.4^\circ$ (c 0.5, CHCl_3).
12. The diol $\beta\text{-D-8}$, when treated with an excess of $\text{CH}_2=\text{CHCH}_2\text{Br}$ and KOH in THF, gave the diallyl ether $\beta\text{-D-9}$, m.p. 204-206°, $[\alpha]_{\text{D}} - 62.2^\circ$ (c 0.5, CHCl_3) in 70% yield. Ozonolysis of $\beta\text{-D-9}$ in methanol, followed by borohydride reduction, afforded (80%) $\beta\text{-D-10}$, m.p. 112-114°, $[\alpha]_{\text{D}} - 64.3^\circ$ (c 0.5, CHCl_3).
13. Condensation (NaH/DMSO) of equimolar proportions of $\beta\text{-D-7}$ and D-11 gave $\beta\text{-DD-12}$ in 15% yield after medium pressure liquid chromatography on silica (ether-light petroleum, b.p. 60-80°).
14. ^1H N.m.r. data (CD_2Cl_2) for $\beta\text{-DD-2}$: δ 7.52-7.35 (m, 5H, aromatic protons), 5.51 (s, 1H, benzyldiene CH), 4.30-3.37 (m, 31H, other CH and CH_2 protons; s at 3.53, 3H, OCH_3), and 1.37, 1.36, 1.32, and 1.30 (4 x s, 12H, 4 x CH_3).
15. Condensation (NaH/DMSO) of equimolar proportions of $\beta\text{-D-10}$ and D-11 gave $\beta\text{-DD-4}$ in 43% after chromatography on alumina (ether) and vacuum distillation (b.p. 240° at 0.05 mm Hg).
16. ^1H N.m.r. data (CD_2Cl_2) for $\beta\text{-DD-4}$: δ 7.50-7.30 (m, 5H, aromatic protons), 5.53 (s, 1H, benzyldiene CH), 4.37-3.05 (m, 31H, other CH and CH_2 protons; s at 3.50, 3H, OCH_3), and 3.37 and 3.31 (2xs, 12H, 4x CH_3).
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20. The acetal group associated with the anomeric centre in the $\beta\text{-D-3-}\alpha\text{-13.HClO}_4$ complex is sterically more accessible to the phenyl ring of the $\text{PhCH}_2\text{NH}_3^+$ ion than is the acetal group of the 1,3-dioxan ring in the $\beta\text{-D-3-}\beta\text{-13.HClO}_4$ complex.

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